

RIVER VALLEY HIGH SCHOOL JC2 PRELIMINARY EXAM

H2 PHYSICS 9749 PAPER 2

13 SEP 2021

2 HOURS

CONFIDENTIAL

CANDIDATE NAME				 		 	
CENTRE NUMBER	S				INDEX NUMBER		
CLASS	2	0	J				

INSTRUCTIONS TO CANDIDATES

DO NOT OPEN THIS BOOKLET UNTIL YOU ARE TOLD TO DO SO.

Read these notes carefully.

Write your name, centre number, index number and class in the spaces at the top of this page.

Write in dark blue or black pen on both sides of the paper. You may use an HB pencil for any diagrams or graphs. Do not use staples, paper clips, glue or correction fluid.

The use of an approved scientific calculator is expected where appropriate.

Candidates answer on the Question Paper. No Additional Materials are required.

Answer all questions.

The number of marks is given in brackets [] at the end of each question or part question.

FOR EXAMINERS' USE		
1	17	
2	/ 10	
3	/ 15	
4	/ 10	
5	/ 5	
6	/8	
7	/ 5	
8	/ 20	
Deduction		
TOTAL	/ 80	

This document consists of 23 printed pages.

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Data

speed	of light	in free	space,
-------	----------	---------	--------

$$c = 3.00 \times 10^8 \text{ m s}^{-1}$$

$$\mu_{\rm O}$$
 = $4\pi \times 10^{-7} \, \rm H \, m^{-1}$

$$\varepsilon_{\rm O}$$
 = 8.85 × 10⁻¹² F m⁻¹

$$(1/(36\pi)) \times 10^{-9} \text{ F m}^{-1}$$

$$e = 1.60 \times 10^{-19} \, \text{C}$$

$$h = 6.63 \times 10^{-34} \,\mathrm{J s}$$

$$u = 1.66 \times 10^{-27} \text{ kg}$$

$$m_e = 9.11 \times 10^{-31} \text{ kg}$$

$$m_{\rm p} = 1.67 \times 10^{-27} \, \rm kg$$

$$R = 8.31 \text{ J K}^{-1} \text{ mol}^{-1}$$

$$N_A = 6.02 \times 10^{23} \text{ mol}^{-1}$$

$$k = 1.38 \times 10^{-23} \text{ J K}^{-1}$$

$$G = 6.67 \times 10^{-11} \text{ N m}^2 \text{ kg}^{-2}$$

$$q = 9.81 \,\mathrm{m \, s^{-2}}$$

Formulae

uniformly	accelerated	motion
uninoming	accici atca	111001011

$$s = ut + \frac{1}{2}at^2$$

$$v^2 = u^2 + 2as$$

$$W = p\Delta V$$

$$p = \rho g h$$

$$\phi = -Gm/r$$

$$p = \frac{1}{3} \frac{Nm}{V} < c^2 >$$

 $T/K = T/^{\circ}C + 273.15$

mean translational kinetic energy of an ideal gas molecule

$$E = \frac{3}{2}kT$$

displacement of particle in s.h.m.

$$x = x_0 \sin \omega t$$

$$v = v_0 \cos \omega t$$
$$= \pm \omega \sqrt{{x_0}^2 - x^2}$$

electric current
$$I = Anvq$$

$$I = Anvq$$

$$R = R_1 + R_2 + \dots$$

$$1/R = 1/R_1 + 1/R_2 + ...$$

$$V = \frac{Q}{4\pi\varepsilon_0 r}$$

$$x = x_0 \sin \omega t$$

$$B = \frac{\mu_0 I}{2\pi d}$$

$$B = \frac{\mu_0 NI}{2r}$$

magnetic flux density due to a long solenoid

$$B = \mu_0 nI$$

$$x = x_0 \exp(-\lambda t)$$

$$\lambda = \frac{\ln 2}{t_{\frac{1}{2}}}$$

Answers all questions in the spaces provided.

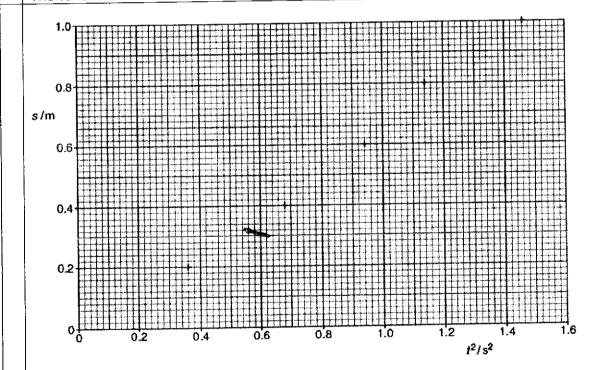
A student takes measurements to determine the acceleration of a ball as it rolls down a slope. He uses the apparatus illustrated in Fig. 1.1.

Fig. 1.1

steel ball

The student measures the time *t* for the ball to roll a distance *s* down the slope after the ball has been released from the electromagnet.

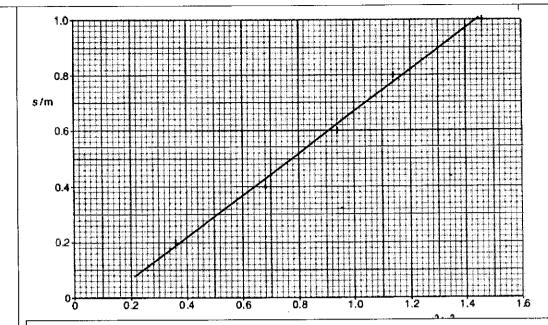
The variation with t^2 of the distance is shown in Fig. 1.2.



(i)

Fig. 1.2

Use Fig. 1.2 to determine a value for the acceleration of the ball down the slope.



RÉL.

Using $s = ut + \frac{1}{2}at^2$ and initial velocity is zero, the gradient of graph is $\frac{1}{2}a$.

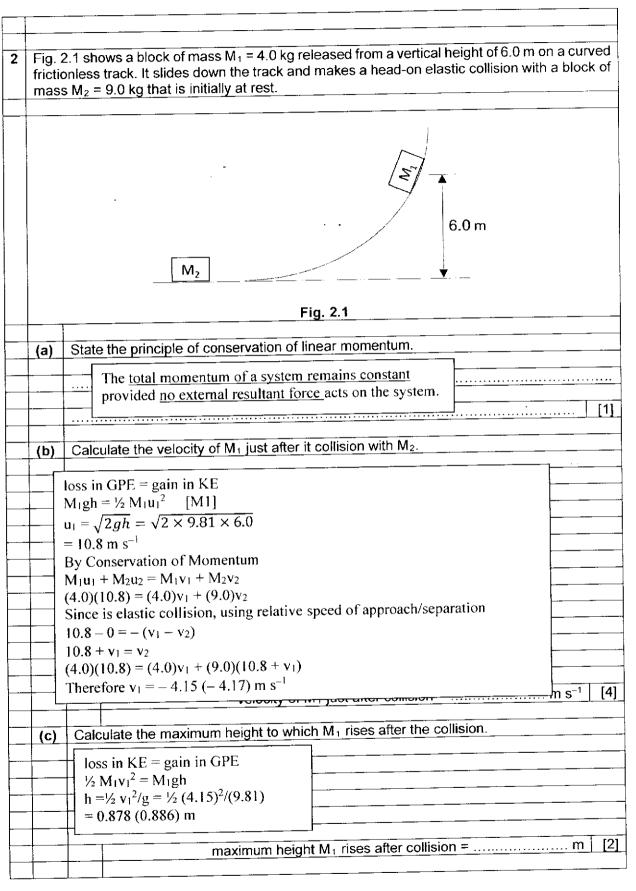
Using (0.40, 0.22) and (1.20, 0.82) to find gradient,

$$gradient = \frac{0.82 - 0.22}{1.20 - 0.40} = 0.75 \text{ [M1] } (0.73 \rightarrow 0.77)$$

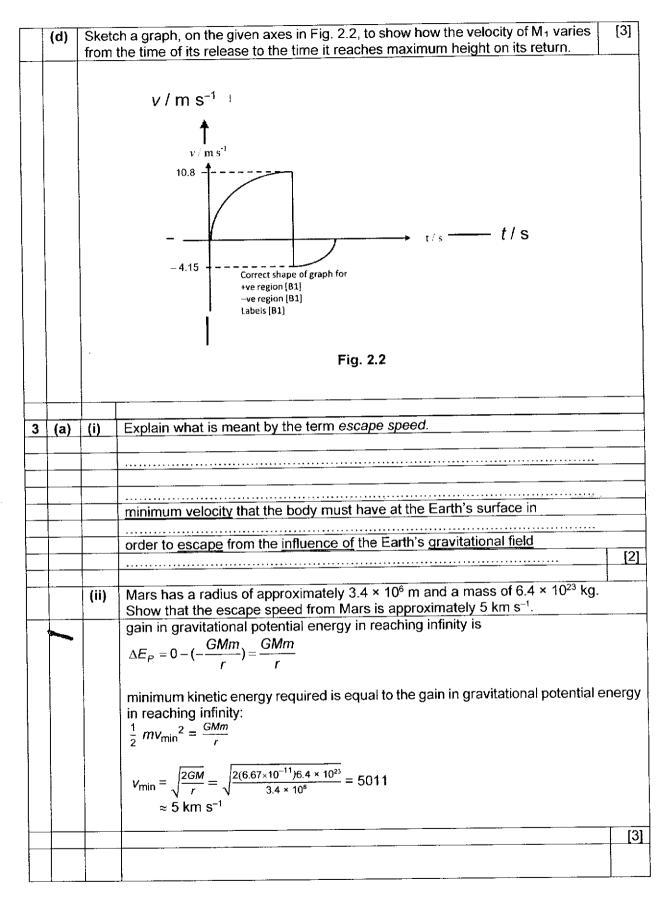
Hence, acceleration = $0.75 \times 2 = 1.50 \ ms^{-2}$.

		acceleration =m s ⁻²
	l	
(ii)	Stat	e the feature of the data shown in Fig. 1.2 that indicates the presence of
	1.	random error,
		The data points are scattered about the line of best fit.
-	2.	and systematic error.
		The best-fit line does not pass through the origin since the
	-	ball has zero displacement when the ball starts to move.

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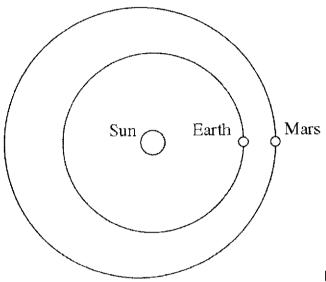


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(iii)	Suggest why a rocket would be able to escape from Mars with an initial speed much less than the escape speed given in part (a)(ii).
	can be done at any speed given sufficient fuel
	OR differentiates from escape speed – unpowered etc \ OR fuel/chemical energy

(b) Fig. 3.1 shows the Sun, Earth and Mars in alignment. Earth and Mars rotate around the Sun in the same directional sense.



Not to scale

Fig. 3.1

A rocket of mass 2.05×10^6 kg leaves the surface of Mars closest to Earth and heads for Earth.

Fig. 3.2 below gives data relevant to the rocket at the start of its journey.

astronomical object (AO)	mass of AO / kg	distance of rocket from the centre of AO / m	rocket's gravitational potential due to AO / J kg ⁻¹	sign of gravitational potential
Mars	6.4×10^{23}	3.4 × 10 ⁶	1.26 × 10 ⁷	
Earth	6.0 × 10 ²⁴	5.6 × 10 ¹⁰		negative
Sun	2.0 × 10 ³⁰	2.3 × 10 ¹¹	5.80 × 10 ⁶	

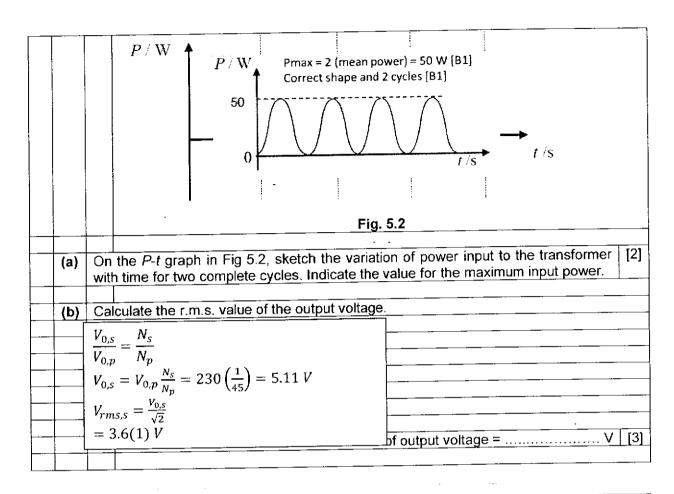
Fig. 3.2

	(i)	Complete Fig. 3.2 by calculating the gravitational potential of the rocket due to the presence of Earth and the signs of the gravitational potential energies due to Mars itself and the Sun.
		$\phi = -\frac{GM}{r} = -\frac{\left(6.67 \times 10^{-11}\right)\left(6.0 \times 10^{24}\right)}{5.6 \times 10^{10}}$ $= -7.15 \times 10^{3} \text{ J kg}^{-1}$
		(formula provided hence no mark)
		negative, negative
		[2]
	(ii)	Calculate the total gravitational potential energy of the rocket on the surface of Mars. As rocket is at surface of Mars, i.e. start of journey, and there is potential due to all 3 Aos, thus
		total GPE of rocket = m($\Sigma \phi$) = 2.05 × 10 ⁶ (-5.80 × 10 ⁶ + -1.26 × 10 ⁷ + -7.15 × 10 ³) = -3.77 × 10 ¹³ (ecf from table)
		total gravitational potential energy =
(c)	(i)	Derive an expression to show that for satellites in a circular orbit
		$\mathcal{T}^2 \propto r^3$
		where T is the period of orbit and r is the radius of the orbit.
		LO9c gravitational force provides centripetal force
		$\frac{GM_sm}{r^2} = mr\omega^2 = mr\left(\frac{2\pi}{T}\right)^2$ $\Rightarrow T^2 = \frac{4\pi^2}{GM_s}r^3$
		$\Rightarrow T^2 \propto T^3$
	(ii)	The orbits of the Earth and Mars can be approximated to be circular orbits around the Sun.
		Hence, estimate the orbital period of Mars.
		$T^2 \propto r^3$
		$\frac{T_{Mars}^2}{365^2} = \frac{\left(2.3 \times 10^{11} + 3.4 \times 10^6\right)^3}{\left(2.3 \times 10^{11} - 5.6 \times 10^{10}\right)^3}$
		,
		$T = 554 \text{ days} \approx 600 \text{ days (1 s.f.)}$
		orbital period of Mars = days [3]

4	The g	ed mass of ideal gas at a low temperature is trapped in a container at constant pressure. gas is then heated and the volume of the container changes so that the pressure stays at × 10 ⁵ Pa.						
		In the gas reaches a temperature of 0.00 °C the volume is 2.20×10^{-3} m ³ .						
	(a)	Draw a graph on the axes below to show how the volume of the gas varies with temperature in °C. [2]						
		volume / 10 ³ m ³						
;		0.5						
		-300 -250 -200 -150 -100 -50 0 50						
		graph passes through given point 2.2 × 10 ⁻³ m ³ at 0.00 °C straight line with positive gradient						
		(straight) line to aim or pass through –273 °C at zero volume						
_	(b)	Calculate the number of moles of gas in the container.						
		LO8d n = P V/R T						
		= 1.00 × 10 ⁵ × 2.20 × 10 ⁻³ /(8.31 × 273.15) □						
		= 0.0969 (moles)						
		number of moles =						
	(c)	Calculate the average kinetic energy of a molecule when this gas is at a temperature of						
	\~/	50.0 °C.						
		LO8h mean kinetic energy = 3/2 K T						
		= 3/2 × 1.38 × 10 ⁻²³ × 323.15 □ = 6.69 × 10 ⁻²¹ (J)						
		0.00 10 (0)						

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1		kinetic energy =
	(d)	Hence or otherwise, calculate the total internal energy of the gas at a temperature of 50.0 °C.
		total internal energy = 6.69 × 10 ⁻²¹ × 0.0969 × 6.02 × 10 ²³ = 390 (J)
		internal energy = J [2]
	(e)	By considering the collisions of gas molecules with the walls of the container, explain why the volume of the container must change if the pressure is to remain constant as the temperature increases.
		As mean square speed of molecules is proportional to temperature, as temperature increases so does change of momentum or change in velocity, hence pressure will increase.
		Increasing the volume compensates for longer time between collisions (rate of collision decreases) as the temperature increases, also as the volume increases the surface area increases which reduces the pressure
		[2]
5	A g tran	raph of the voltage input to an ideal transformer is shown in Fig. 5.1. The 45:1 step-down isformer has a mean input power of 25 W.
		V/V
		230
		-230 t/s
_		Fig. 5.1



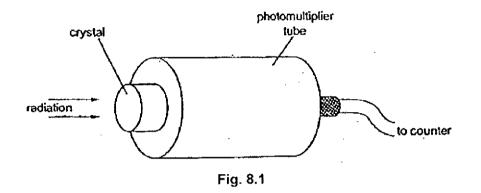
6	(a)	Using energy considerations, distinguish between electromotive force and potential difference.
		The electromotive force (e.m.f.) of a source is defined as the <u>amount of energy transferred</u> from non-electrical forms of energy to electrical energy per unit charge as the charge passes through a complete circuit.
		The potential difference <i>V</i> between two points of a circuit is defined as the amount of electrical energy converted to other forms of energy per unit charge moved between these two points.
_		
		[2]
_	-	

(b)	An e	electr	ic hotplate is designed to operate on a power supply of 240 V and has two c of resistivity of 9.8 \times 10 ⁻⁷ Ω m. Each coil of wire has a length	oils of
	16 n	vire n and I	or resistivity of 9.8 \times 10 $^{\circ}$ Ω III. Each coir of whe has a length discress-sectional area 0.20 mm ² .	- J
	(i)	For	one of the coils, calculate	
	1.0	1.	its resistance,	
		,	R = $(9.8 \times 10^{-7}) (16) / (0.20 \times 10^{-6})$ R = 78.4Ω	
			resistance =Ω	[2]
		2.	the power dissipation when a 240 V supply is connected across it.	
			P = (240) ² / 78.4 P = 735 W	
		<u> </u>	power = W	[2]
	(ii)	Or	Fig. 6.2, fill up the table with "ON" or "OFF" to obtain the lowest and highest rels of operating power. Switch A	
			Fig. 6.2	[2]
				,-J
Diver Vall			switch A switch B switch C Pa 13 of 21 H2 Physics 97	L

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									$\overline{}$
				Lowest	OFF	ON	OFF		
				Highest	ON	OFF	ON		
			B1 for each Lo	owest and F	lighest row				
7	Fig	7 1 sl	nows a microwa	ve transmit	ter aimed at	a metal plat	e		
•	1 19.	1.10	10110 4 1111010111						
		tr	ansmitter	X • m	icrowave	Y ●··-··• detector	metal p	direction of motion	
							metal p	late	
					Fig. 7	7.1			
	(a)	A sr 12 th	mall detector, n	noved along dal position	the line XY	travels 16	cm in movin	g from the fi	rst to the
		Sho	w that the micro	owaves emi	tted have a f	requency of	10 GHz.		_[3]
+	Hence There = (3 >	e 11 > fore t < 10 ⁸)	the 1 st and 12 th not $\lambda/2 = 16$ cm $\frac{\lambda}{2}$ frequency of the $\frac{\lambda}{2}$ (0.0291) $\frac{\lambda}{2}$ Hz ~ 10 GHz	$\lambda = 0.0291$ microwave	l m		V2)		
	+-								
	(b)		detector is no ction XY at a co			metal plate	e is moved t	to the right, a	along the
		Stat	te and explain v	vhat is obse	erved at the o	letector.			
	4-4	The d	letector will con path/phase differ letal plate shifts	ntinue to <u>det</u> rence betwe	ect a series cen the incide	f maxima ar	nd minima. cted waves c	changes as	
	-								
									[2]
	1							 	

During radioactive decay, which is spontaneous and random, γ -ray (gamma ray) photons may be emitted. When these photons are incident on a sodium iodide crystal, some of the photons may be absorbed in the crystal. The absorption of a γ -ray photon causes the emission of a short pulse of light known as a scintillation. The scintillations may be detected and converted into electrical pulses using a photomultiplier tube, which, when connected to a counter, gives the count rate and enables γ -ray activity to be measured. The arrangement is illustrated in Fig. 8.1.



The crystals used in such counters may be of different shapes. Fig. 8.2 shows a solid cylindrical crystal.

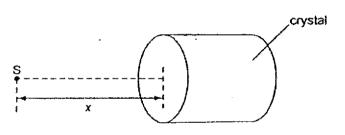


Fig. 8.2

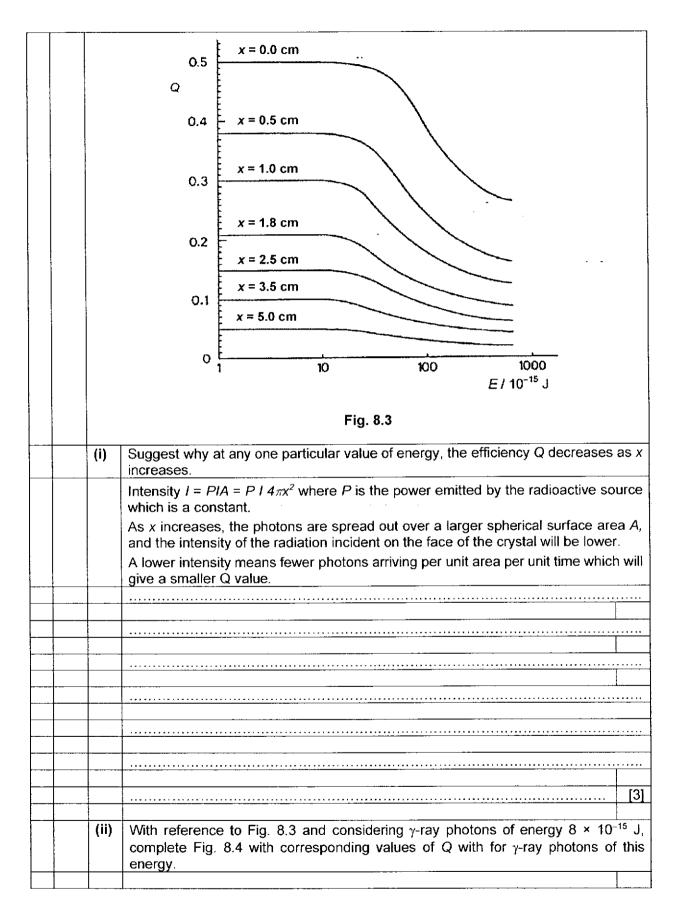
The small γ -ray source S is placed a distance x in front of one face of the crystal. The source is assumed to emit photons uniformly in all directions. Not all of the emitted photons will be absorbed by the crystal. The efficiency Q of detection is defined by the equation

Q = number of photons producing scintillations in the crystal total number of photons emitted by the source

(a)	By reference to the passage, explain what is meant by			
	(i)	count rate		
		Count rate is a measure of the rate of radiation received by a radioactivity detector		

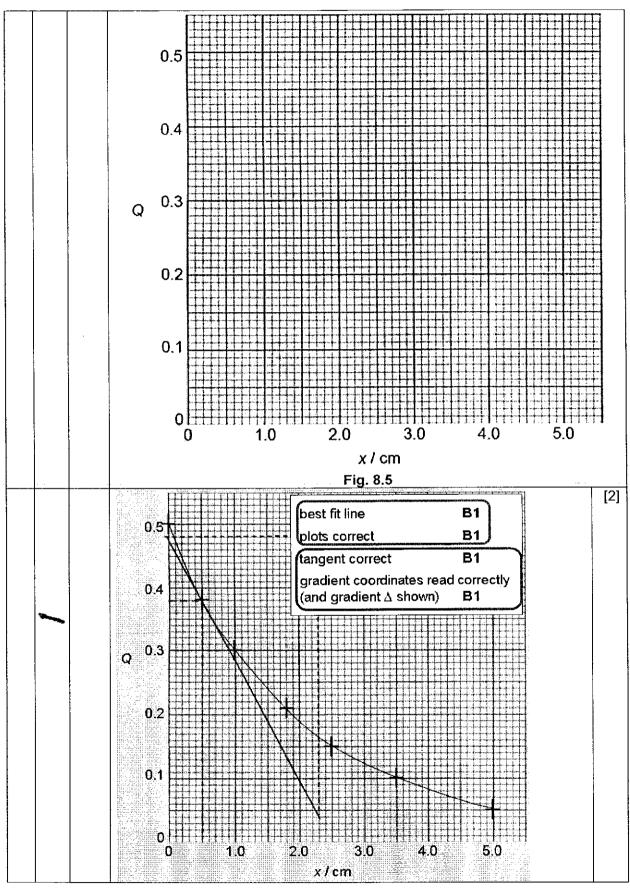
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		[1]
		_[']
(ii)	Activity of a radioactive isotope is defined as the number of nuclear disintegrations and times	ions
	per unit time.	<u>.,</u>
		_[′
		ed i
	The source emits photons in all directions and many will travel in directions awa from the face of the crystal and not be absorbed.	
	Only those photons that arrive at the small circular window area of the face of t crystal will have a probability of being absorbed.	he
1.		
]_[
	γ -rays are highly penetrating hence not every γ -ray photon that enters the crystal interact with a sodium iodide molecule.	
i	Some of the higher energy photons will penetrate through the crystal without interaction taking place.	. aı
2.		
		<u>:</u>
		<u></u>
	Suggethe of	Activity of a radioactive isotope is defined as the number of nuclear disintegrative per unit time. Suggest two reasons why the γ-ray photons emitted by the source are not all absorbed the crystal. The source emits photons in all directions and many will travel in directions away from the face of the crystal and not be absorbed. Only those photons that arrive at the small circular window area of the face of the crystal will have a probability of being absorbed. 1. γ-rays are highly penetrating hence not every γ-ray photon that enters the crystal interact with a sodium iodide molecule. Some of the higher energy photons will penetrate through the crystal without interaction taking place.



	T		x / cm	
		Q	X / GIII]
				1 1
				1
				-
	<u> </u>			-
+]
				[2]
		Fig. 8	.4	[2]
	Q	x / cm		
	0.05	5.0		
	0.10	3.5		
	0.15	2.5		
	0.21	1.8		
	0.30	1.0		
	0.38	0.5		
	0.50	0.0		

	(d) (i)	Use your values in Fig. 8.4 to draw a graph of Q against x in Fig. 8.5.	
-	i	ĺ		J



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(i	ii)	Use Fig. 8.5 to determine the rate of change of Q with x when $x = 0.5$ cm.	
		Gradient = ∆y/∆x -0.19 to -0.21 cm ⁻¹	
		rate of change = cm ⁻¹	[4
	iii)	It may be deduced from Fig. 8.5 that Q is related to x by an expression of the	forr
		Q = ae ^{-bx}	
		where a and b are constants.	
		Explain how Fig. 8.5 shows this. The equation of the form $Q = ae^{-bx}$ is an exponentially decreasing function	
		with respect to x.	
		with respect to x. An exponentially decreasing function, similar to the one for radioactive decay with respect to time, will fall to half its value for every fixed interval.	
		with respect to x. An exponentially decreasing function, similar to the one for radioactive	
		with respect to <i>x</i> . An exponentially decreasing function, similar to the one for radioactive decay with respect to time, will fall to half its value for every fixed interval. It can be seen from the graph that this fixed interval is approximately 1.5	
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(e)	(i)	Suggest why the maximum efficiency that can be achieved with low-energy γ -ray photons using the crystal illustrated in Fig. 8.2 is 0.50.
		The maximum efficiency is achieved when $x = 0$ cm, i.e. the source is placed right at the face of the crystal.
		In this case, half the γ -ray photons emitted (those to the right side of the source within a hemispherical surface area) will be captured by the crystal.
 -		

	[2]
- (ii)	A second crystal consists of a hollow cylinder, as shown in Fig. 8.6. Fig. 8.6
	State the effect of this change of shape on the maximum efficiency Q of the detector if the source is placed inside the hollow region.
	The maximum efficiency Q will be greater as the available area for capturing the γ-ray photons is larger.
	[1]

END OF PAPER